

Synthesis of 1,3-Dioxin-4-ones having a Trifluoromethyl Group at the 5-Position: Versatile Building Blocks for Trifluoromethylated Compounds

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The reaction of 5-iodo-1,3-dioxin-4-ones with trifluoromethyl iodide in the presence of copper powder in hexamethylphosphoric triamide affords the title compounds, which can be converted either to α -trifluoromethylated β -keto acid derivatives or to uracil or oxazine derivatives bearing a trifluoromethyl group.

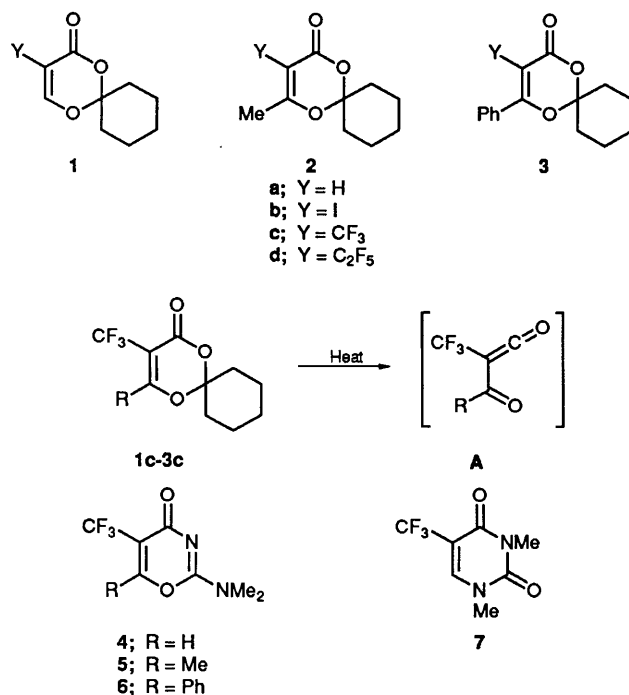
We have recently synthesized 5-fluoro-1,3-dioxin-4-ones and utilized them as building blocks for monofluorinated derivatives either of formylacetic acid or of heterocyclic systems.¹ Since introduction of trifluoromethyl groups into a variety of organic molecules is important, as is introduction of fluorine² in fluorine chemistry, much effort has been expended in elaboration of trifluoromethylation methods.³ We now report the synthesis of 5-trifluoromethyl-1,3-dioxin-4-ones and some of their reactions.

When the 5-iodo-1,3-dioxin-4-ones **1b**, **2b** and **3b**, synthesized from the 5-unsubstituted dioxinones **1a**, **2a** and **3a** by our procedure,^{1,4} were allowed to react with trifluoromethylcopper in hexamethylphosphoric triamide⁵ (ca. 50 °C), the corresponding trifluoromethylated derivatives **1c** (m.p. 77.5–78.0 °C) **2c** (oil) and **3c** (m.p. 160–161 °C)^{†‡} were obtained in satisfactory yields (ca. 65%). Small amounts (ca. 5–10%) of the pentafluoroethyl derivatives **1d** (oil) and **2d** (m.p. 64.0–64.5 °C) as well as the 5-unsubstituted derivative **1a** were obtained concomitantly. It is obvious that the former compounds have been formed by coupling of the pentafluoroethyl complex generated from the trifluoromethyl complex⁶ and **1a** was formed by reduction with copper powder present in the reagent.⁷

The following transformations using **1c**, **2c** and **3c** have demonstrated that these dioxinones serve *via* the ring-opened ketene species (**A**) either as the equivalents of β -keto acids or the precursors of uracils and oxazinones having a trifluoromethyl group. Thus, reaction of **2c** with dimethylcyanamide in

refluxing xylene gave the expected oxazinone **5** (m.p. 90–91.5 °C)[‡] in 81% yield.[§] In the same manner, the oxazinone **4** (m.p. 113–114 °C) and **6** (m.p. 185–186 °C) were obtained from **1c** and **3c**, respectively. The reaction of **1c** with 1,3-dimethylurea in refluxing toluene led to the corresponding uracil **7** (m.p. 108–109 °C)^{‡§} in 84% yield.

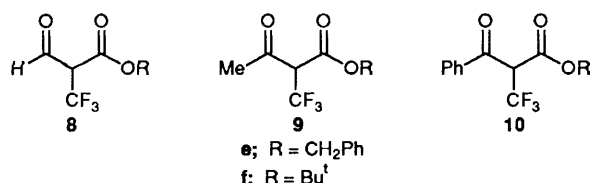
The introduction of a trifluoromethyl group at an aliphatic carbon atom is more difficult than bonding trifluoromethyl to



[†] All new compounds exhibited satisfactory spectroscopic (NMR and IR) and combustion or high resolution mass spectral analytical data.

[‡] **1c**: ¹H NMR (CDCl₃) δ : 1.2–2.3 (10H, m, 5 \times CH₂), 7.70 (1H, s, 6-H). **2c**: ¹H NMR (CDCl₃) δ : 1.2–2.3 (10H, m, 5 \times CH₂), 2.27 (3H, q, *J* 2 Hz, 6-H). **3c**: ¹H NMR (CDCl₃) δ : 1.2–2.3 (10H, m, 5 \times CH₂), 7.55 (5H, s, Ph). **5**: ¹H NMR (CDCl₃) δ : 2.40 (3H, q, *J* 2 Hz, CH₃), 3.17 (6H, s, 2 \times NCH₃). **7**: ¹H NMR (CDCl₃) δ : 3.37 (3H, s, CH₃), 3.50 (3H, s, CH₃), 7.70 (1H, s, 6-H). **9e** (keto/enol = ca. 9): ¹H NMR (CDCl₃) δ : 2.30 (3H, s, CH₃), 4.23 (1H \times 0.9, q, *J* 8 Hz, 2-H), 5.27 (2H, s, CH₂), 7.37 (5H, s, Ph), 14.00 (1H \times 0.1, s, OH).

[§] This paper reports reactions using dioxinones having a spiro-linked pentamethylene chain at the 2-position; reactions of the corresponding 2,2-dimethyl derivatives proceeded similarly.



an aromatic nucleus. The synthesis of α -trifluoromethylated β -keto esters is particularly difficult, because, under basic conditions, these α -trifluoromethylated compounds are readily dehydrofluorinated to give β,β -difluoroacrylic acid intermediates which are prone to hydrolysis and decarboxylation.⁹ Though several methods⁹⁻¹¹ are available for the synthesis of these esters, there is a clear need for a new and general method. We have developed a simple one-pot synthesis of β -keto esters containing a trifluoromethyl group in the α -position using the dioxinones **1c**, **2c** and **3c** as the starting materials. Thus, according to the general procedure for the conversion of the dioxinones to β -keto esters,¹² **2c** was refluxed in xylene containing benzyl alcohol to give the ester **9e** (oil)[‡] in nearly quantitative yield. This reaction proceeded, irrespective of either the 6-substituent in the dioxinones (**1-3**) or the alcohols (PhCH₂OH or Bu^tOH) to give the corresponding β -keto esters **8e**, **f**, **9e**, **f**, and **10e**, **f**.

We are now studying the use of **1c**, **2c** and **3c** in pericyclic (Diels-Alder¹³ and photo[2 + 2]cycloaddition¹⁴) reactions,¹⁵ in which the C-C double bond in the dioxinone ring is used without destroying the ring.

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References

- M. Sato, C. Kaneko, T. Iwaoka, Y. Kobayashi and T. Iida, *J. Chem. Soc., Chem. Commun.*, 1991, 699.
- The synthesis of fluorinated organic compounds as well as their novel physical and biological properties have been reviewed: *Biochemical Aspects of Fluorine Chemistry*, eds. R. Filler and Y. Kobayashi, Kodansha Ltd, Tokyo, 1982; J. T. Welch, *Tetrahedron*, 1987, **43**, 3125; T. B. Patrick, *J. Chem. Ed.*, 1979, **56**, 228.
- I. Kumadaki, *Yakugaku Zasshi*, 1985, **105**, 713.
- M. Sato, N. Yoneda and C. Kaneko, *Chem. Pharm. Bull.*, 1986, **34**, 4577.
- The most useful trifluoromethylation reaction employs organo-metallic reagents, trifluoromethyl-copper or -zinc complexes and has wide applicability for aryl halides and related compounds: Y. Kobayashi, K. Yamamoto and I. Kumadaki, *Tetrahedron Lett.*, 1979, 4071; T. Kitazume and N. Ishikawa, *Chem. Lett.*, 1982, 137.
- The disproportionation of [CF₃Cu] to [CF₃CF₂Cu] was demonstrated by a spectroscopic study and by the formation of pentafluoroethylated products in trifluoromethylations using trifluoromethylcopper complexes: D. W. Wiemers and D. J. Burton, *J. Am. Chem. Soc.*, 1986, **108**, 832; Y. Kobayashi and I. Kumadaki, *J. Chem. Soc., Perkin Trans. 1*, 1980, 661.
- Y. Kobayashi, K. Yamamoto, T. Asai, M. Nakano and I. Kumadaki, *J. Chem. Soc., Perkin Trans. 1*, 1980, 2755.
- K. Fuchikawa and I. Ojima, *Tetrahedron Lett.*, 1982, **23**, 4099; see also T. Fuchikami, A. Yamanouchi and I. Ojima, *Synthesis*, 1984, 766.
- T. Yokozawa, T. Nakai and N. Ishikawa, *Tetrahedron Lett.*, 1984, **25**, 3987; T. Stephan, S. T. Purrington and C. L. Bumgardner, *J. Org. Chem.*, 1984, **49**, 3702 and references cited therein.
- K. Uneyama and K. Ueda, *Chem. Lett.*, 1988, 853.
- T. Yokozawa, T. Nakai and N. Ishikawa, *Tetrahedron Lett.*, 1984, **25**, 3991.
- Synthesis of β -keto esters from dioxinones via ketene trapping has been reported: M. Sato, N. Yoneda, N. Katagiri, H. Watanabe and C. Kaneko, *Synthesis*, 1986, 672.
- M. Sato, Y. Abe and C. Kaneko, *J. Chem. Soc., Perkin Trans. 1*, 1991, 1779.
- M. Sato, Y. Abe, K. Takayama and C. Kaneko, *J. Heterocycl. Chem.*, 1991, **28**, 241 and references therein.
- A variety of reactions using dioxinones have been reviewed: C. Kaneko, M. Sato, J. Sakaki and Y. Abe, *J. Heterocycl. Chem.*, 1990, **27**, 25.